Mechanisms for a nutrient-conserving carbon pump in a seasonally stratified, temperate continental shelf sea

4	Matthew P. Humphreys ^{1,2,*} , Eric P. Achterberg ^{1,3} , Joanne E. Hopkins ⁴ , Mohammed Z. H.
5	Chowdhury ¹ , Alex M. Griffiths ¹ , Susan E. Hartman ⁵ , Tom Hull ^{2,6} , Angelina Smilenova ¹ ,

- 6 Juliane U. Wihsgott⁷, E. Malcolm S. Woodward⁸, C. Mark Moore¹
- 7
- 8 ¹ Ocean and Earth Science, University of Southampton, Southampton, UK
- 9 ² Centre for Ocean and Atmospheric Sciences, School of Environmental Sciences, University
- 10 of East Anglia, Norwich, UK
- ³ GEOMAR Helmholtz Centre for Ocean Research, Kiel, Germany
- ⁴ National Oceanography Centre, Liverpool, UK
- ⁵ National Oceanography Centre, Southampton, UK
- ⁶ Centre for Environment, Fisheries and Aquaculture Science, Lowestoft, UK
- ⁷ Department of Earth, Ocean and Ecological Sciences, School of Environmental Sciences,
- 16 University of Liverpool, UK
- ⁸ Plymouth Marine Laboratory, Plymouth, UK
- 18
- 19 * Corresponding author (<u>matthew.humphreys@uea.ac.uk</u>)
- 20

21 Abstract

22 Continental shelf seas may have a significant role in oceanic uptake and storage of carbon 23 dioxide (CO₂) from the atmosphere, through a 'continental shelf pump' mechanism. The 24 northwest European continental shelf, in particular the Celtic Sea (50°N 8°W), was the target 25 of extensive biogeochemical sampling from March 2014 to September 2015, as part of the 26 UK Shelf Sea Biogeochemistry research programme (UK-SSB). Here, we use the UK-SSB 27 carbonate chemistry and macronutrient measurements to investigate the biogeochemical 28 seasonality in this temperate, seasonally stratified system. Following the onset of 29 stratification, near-surface biological primary production during spring and summer removed 30 dissolved inorganic carbon and nutrients, and a fraction of the sinking particulate organic 31 matter was subsequently remineralised beneath the thermocline. Water column inventories of 32 these variables throughout 1.5 seasonal cycles, corrected for air-sea CO₂ exchange and 33 sedimentary denitrification and anammox, isolated the combined effect of net community 34 production (NCP) and remineralisation on the inorganic macronutrient inventories. Overall 35 inorganic inventory changes suggested that a significant fraction (>50%) of overall NCP of around 3 mol-C m⁻² yr⁻¹ appeared to be stored within a long-lived organic matter (OM) pool 36 37 with a lifetime of several months or more. Moreover, transfers into and out of this pool 38 appeared not to be in steady state over the one full seasonal cycle sampled. Accumulation of 39 such a long-lived and potentially C-rich OM pool is suggested to be at least partially responsible for the estimated net air-to-sea CO₂ flux of ~1.3 mol-C m⁻² yr⁻¹ at our study site, 40 41 while providing a mechanism through which a nutrient-conserving continental shelf pump for 42 CO₂ could potentially operate in this and other similar regions.

44 **1. Introduction**

45 The ocean is an important buffer for atmospheric carbon dioxide (CO₂), naturally storing a 46 large pool of dissolved inorganic carbon (C_T), and also absorbing about a quarter of annual

- 47 anthropogenic CO_2 emissions (Le Quéré et al., 2009) and thereby mitigating their climatic
- 48 impacts (IPCC, 2013). The timescale on which the ocean's surface mixed layer equilibrates
- 49 CO_2 with the atmosphere varies spatially, but is typically from a few months to a year
- 50 (Broecker and Peng, 1974; Jones et al., 2014). Transfer from the surface to the deep interior
- 51 is a much slower rate-limiting step, which is performed by the solubility, soft tissue,
- 52 carbonate and mixed-layer 'pumps' in the open ocean (Volk and Hoffert, 1985; Dall'Olmo et
- al., 2016). Tsunogai et al. (1999) first proposed the existence of an additional 'continental
- shelf pump' (CSP) that facilitates $C_{\rm T}$ transfer from shallow coastal surface waters into the
- 55 interior ocean, and thereby enhances local oceanic CO₂ uptake. The CSP operates through
- 56 autotrophic conversion of $C_{\rm T}$ into organic matter (OM) in a continental shelf sea, which
- 57 drives a compensatory net air-to-sea CO₂ flux (Chen and Borges, 2009; Chen et al., 2013;
- 58 Laruelle et al., 2014). In order to prevent the fixed CO₂ from being quickly released back into
- 59 the atmosphere, the carbon-enriched shelf waters need to be transported into the open ocean,
- 60 beneath the seasonal thermocline (Tsunogai et al., 1999; Thomas et al., 2004).

61 The Celtic Sea is a seasonally stratified, temperate sea that forms part of the northwest 62 European continental shelf (Fig. 1). Previous studies of its surface waters have shown that it 63 acts as a net sink of atmospheric CO₂ (Frankignoulle and Borges, 2001; Marrec et al., 2015), 64 which is typical for this type of sea (Borges, 2005; Laruelle et al., 2014). There must 65 therefore be a net flux of carbon (C) out of the shelf sea water column, for example through 66 OM export to sediments, or advective exchange with the open ocean. Celtic Sea sediments 67 are unlikely to form an important component of this C sink, as they are predominantly sandy 68 material with relatively low organic matter accumulation (de Haas et al., 2002; Suykens et al., 69 2011; Diesing et al., 2017), but physical mechanisms that could transport shelf waters into the 70 open ocean have been identified here (Cooper and Vaux, 1949; Shapiro et al., 2003; Ivanov et 71 al., 2004) and in other similar shelf environments (Álvarez-Salgado et al., 2001). In either 72 case, the question arises of how the nutrient supply is sustained that drives the net C uptake. 73 As essential nutrients such as dissolved inorganic nitrogen and soluble reactive phosphorus 74 (DIN and SRP) are also incorporated into OM, but not replenished through air-sea exchange, 75 the net removal of C should be accompanied by a corresponding loss of N and P from the 76 shelf waters (Thomas et al., 2004; Bozec et al., 2006). Further N may be lost from the shelf

77 waters through denitrification and/or anammox in the shelf sediments (Hydes et al., 2004; 78 Kitidis et al., 2017). Thus, in order to sustain the shelf sea nutrient inventories in the presence 79 of net CO₂ uptake from the atmosphere, any incoming nutrient supply would therefore need 80 to be accompanied by a stoichiometric deficit of carbon relative to outgoing waters. The 81 mechanism by which this occurs remains uncertain. One possibility is that riverine inputs 82 provide high-nutrient, low-C waters to the shelf sea. There are also high-nutrient waters in the 83 Irish Sea to the northeast of our study region (Gowen et al., 2002). However, these influences 84 are not thought to be substantial for the Celtic Sea, as on-shelf lateral circulation is relatively 85 slow (Pingree and Le Cann, 1989), and most of the riverine nutrient inputs are quickly 86 removed by biogeochemical processes close to land (Ruiz-Castillo et al., 2017, this issue). In 87 general, riverine nutrient inputs to shelf seas are small relative to inputs at the continental 88 margin, for example through upwelling and internal waves at the shelf break (Wollast, 1998). 89 Also, rivers flowing into the Celtic Sea (e.g. from southern Ireland) typically have high total 90 alkalinity (A_T) due to chalk and limestone bedrock in their catchments (McGrath et al., 2015). 91 This high $A_{\rm T}$ supports high $C_{\rm T}$ concentrations, so the required C deficit relative to N and P is 92 unlikely to occur in these waters.

93 Here, we investigate the hypothesis that OM storage coupled with differences in the 94 stoichiometry of OM remineralisation relative to its production could provide the 95 stoichiometric inconsistency required to sustain the nutrient supply in the Celtic Sea, as has 96 been suggested to occur in similar shelf seas (e.g. Álvarez-Salgado et al., 2001; Bozec et al., 97 2006). This would require the stoichiometric C:N:P ratio for remineralisation to have a lower 98 C coefficient than in the equivalent ratio for net community production (NCP). The shelf 99 nutrient inventory could therefore be sustained through some combination of enhanced 100 remineralisation of non-C nutrients, or the build-up of a stock of C-rich OM that could be 101 transported into the adjacent open ocean prior to complete remineralisation, thereby acting as 102 a nutrient-conserving CO₂ sink. Our approach is to investigate the distributions and 103 inventories of $C_{\rm T}$ and dissolved inorganic nutrients, which record the integrated effects of 104 NCP and OM remineralisation, along with other processes such as air-sea CO_2 exchange. We 105 use new seasonally resolved $C_{\rm T}$ and nutrient observations in this way to estimate the annual 106 magnitude and stoichiometry of NCP and remineralisation at a site in the central Celtic Sea. 107 We subsequently infer seasonal and interannual changes in OM inventories and 108 stoichiometry, and consider mechanisms that could support C export while conserving 109 nutrients, and therefore a CSP.

111 **2. Methods**

112 **2.1. Research cruises**

113 Seawater samples were collected during 10 research cruises to the Celtic Sea on the UK

114 research ships RRS *Discovery* and RRS *James Cook*, from March 2014 to August 2015

115 (Supp. Table S1). All of our datasets are freely available from the British Oceanographic

116 Data Centre.

117 2.1.1. Marine carbonate system

118 Samples for $C_{\rm T}$ and $A_{\rm T}$ were collected via silicone tubing into 250 ml borosilicate glass

119 bottles either from the samplers (Ocean Test Equipment) positioned on a CTD rosette frame,

120 or from the ship's underway seawater supply, following an internationally established

121 protocol (Dickson et al., 2007). Each bottle was sealed shut with a greased ground glass

122 stopper after introducing a 2.5 ml air headspace and sterilising the sample with 50 µl of

123 saturated mercuric chloride solution. All samples were stored in the dark until analysis.

124 Measurements of $C_{\rm T}$ and $A_{\rm T}$ were carried out at the University of Southampton between June

125 2014 and January 2016, using several different instruments. The results were calibrated using

126 measurements of batches 128, 135, 136, 138, 141, 142, 144, 146 and 148 of seawater

127 certified reference material (CRM) obtained from A. G. Dickson at Scripps Institution of

128 Oceanography, San Diego, USA (Dickson et al., 2003).

- 129 Seawater $C_{\rm T}$ was measured using either the Versatile INstrument for the Determination of
- 130 Total inorganic carbon and titration Alkalinity (VINDTA 3C, #024 and #038, Marianda,
- 131 Germany) or the Dissolved Inorganic Carbon Analyzer AS-C3 (Apollo SciTech Inc., USA).
- 132 Both of these instruments first acidify a seawater subsample with excess 10% phosphoric
- 133 acid, to convert all $C_{\rm T}$ into aqueous CO₂. Nitrogen gas is then bubbled through to carry the

134 CO₂ to a detector. In the VINDTA 3C, the detection is by coulometric titration (CM5014 CO₂

- 135 coulometer, UIC Inc., USA), while the AS-C3 uses infrared absorption (LI-7000 CO₂/H₂O
- 136 Analyzer, LI-COR, USA). Based on CRM measurements throughout the entire UK-SSB
- 137 programme, the 1σ precision for $C_{\rm T}$ was $\pm 2.6 \,\mu$ mol kg⁻¹ for the VINDTA 3C (number of
- 138 CRM measurements (*n*) = 547), and \pm 4.0 µmol kg⁻¹ for the Apollo AS-C3 (*n* = 135).
- 139 The A_T was measured using either the VINDTA 3C or the Total Alkalinity Titrator AS-ALK2
- 140 (Apollo SciTech Inc., USA). Both instruments determine A_T by an open-cell, potentiometric
- 141 titration of a seawater subsample with 0.1 M hydrochloric acid. The A_T values from the

- 142 VINDTA 3C measurements were recalculated using a modified Gran plot approach
- 143 (Humphreys, 2015a) as implemented by Calkulate v0.1.2 (freely available from
- 144 <u>https://github.com/mvdh7/calkulate</u>). The CRM measurements through the entire UK-SSB
- 145 programme indicated a 1σ precision for $A_{\rm T}$ of $\pm 2.7 \,\mu$ mol kg⁻¹ for the VINDTA 3C (n = 543),
- 146 and $\pm 3.9 \ \mu\text{mol kg}^{-1}$ for the Apollo AS-ALK2 (n = 109).
- 147 We performed a cross-over analysis using XOVER v1.0.1 (Humphreys, 2015b), which is
- 148 freely available from https://github.com/mvdh7/xover, to assess our $C_{\rm T}$ and $A_{\rm T}$ measurement
- 149 accuracy relative to historical data in GLODAPv2 (Olsen et al., 2016). This used only data
- 150 from deeper than 250 m (i.e. off the shelf, beneath the base of the thermocline, and a
- 151 maximum cross-over distance of 200 km). For $C_{\rm T}$, there was a mean \pm standard deviation
- 152 (SD) offset of $13.6 \pm 7.6 \ \mu mol \ kg^{-1}$ between the combined SSB dataset and nine GLODAPv2
- 153 cruises from 1994 to 2008. Despite the large size of this offset, it does not indicate any
- 154 problem with our measurements, because an ordinary least squares linear regression between
- 155 the offsets and the sampling date reveal a decrease in offset size at a rate of about 1 μ mol kg⁻¹
- 156 yr⁻¹ ($r^2 = 0.66$), corresponding to the anthropogenic C_T increase, and reaching a value of 0
- 157 close to the date of the SSB cruises. For $A_{\rm T}$, the equivalent offset was $6.8 \pm 7.1 \ \mu {\rm mol \ kg^{-1}}$ for
- nine cruises from 1983 to 2008, which is close to the minimum adjustment limit of 6 µmol
- 159 kg⁻¹ for this variable in GLODAPv2 (Olsen et al., 2016). In general, cruises closer in time to
- 160 SSB sampling had a smaller $A_{\rm T}$ offset. We conclude that our measurements are accurate to
- 161 within their precision relative to historical observations. Indeed, the internal consistency of
- 162 data within the GLODAPv2 compilation is reported as ± 4 and $\pm 6 \mu mol kg^{-1}$ for C_T and A_T
- 163 respectively (Olsen et al., 2016).

164 2.1.2. Other variables

165 Macronutrient samples were collected into 60 ml high density polyethylene (HDPE, Nalgene) bottles that had been aged and acid washed before sampling. The measurements (of nitrate, 166 167 nitrite, ammonium, and phosphate) were carried out at sea during the UK-SSB cruises, immediately after sample collection, using the Plymouth Marine Laboratory 5-channel Bran 168 169 and Luebbe AAIII system, following Woodward and Rees (2001). The instrument was 170 calibrated using in-house nutrient standards, and accuracy was monitored by regular 171 comparisons with Nutrient Reference Material obtained from KANSO Technos (Japan). We 172 define dissolved inorganic nitrogen (DIN) as the sum of the concentrations of nitrate, nitrite

- 173 and ammonium, and soluble reactive phosphorus (SRP) as the phosphate concentration. The
- 1σ precisions were $\pm 0.13 \ \mu$ mol kg⁻¹ and $\pm 0.0073 \ \mu$ mol kg⁻¹ for DIN and SRP respectively. 174
- 175 Water column hydrography was measured by Sea-Bird conductivity-temperature-depth
- 176 (CTD) sensors attached to the sampling rosette frame, to determine the temperature (T in °C),
- 177 practical salinity (S) and depth for each discrete biogeochemical sample. The sea surface
- 178 hydrography (T and S) was measured continuously by Sea-Bird sensors through the underway
- 179 seawater supply at a nominal depth of 6 m. The T and S sensor data for both the underway
- 180 and CTD systems were calibrated using a set of discrete samples collected during each cruise.
- We consider the 1σ uncertainty in S to be ± 0.002 (Le Menn, 2011). Further information on 181
- the sensor measurements and processing can be found in the cruise reports. 182

183 2.1.3. Unit conversions

The $C_{\rm T}$ and $A_{\rm T}$ measurements were determined in units of µmol kg⁻¹, while DIN and SRP 184 were reported in mmol m⁻³ under analysis conditions. We therefore converted the latter two 185 into µmol kg⁻¹ assuming a constant analysis temperature of 20°C. For the column inventory 186 analysis, we then converted the $C_{\rm T}$, DIN and SRP measurements from μ mol kg⁻¹ into mmol 187 m⁻³ using the *in situ* density. All densities for these calculations were determined using the 188 Gibbs-SeaWater (GSW) Oceanographic Toolbox v3.05.5 (McDougall and Barker, 2011). 189

190

2.2. Central Celtic Sea mooring

191 A temperature-salinity (TS) chain (a series of sensors positioned vertically along a cable 192 moored to the seafloor) was also deployed on a Cefas SmartBuoy mooring at CCS from 193 March 2014 to July 2015 (Wihsgott et al., 2016). This measured seawater T and S at 194 approximately 2.5 m vertical resolution throughout the water column (i.e. from c. 5 to 145 m) 195 and at 5 minute intervals. A combination of sensor types were used: Star Oddi DST Centi 196 temperature loggers, Star Oddi Starmon mini-underwater temperature recorder and RBRsolo 197 T temperature logger, along with SBE16+ SeaCAT CTD and SBE37 MicroCAT CTD sensors. We calculated mixed layer depth (z_{ml} in m) as the shallowest depth where the 198 potential density increased by 0.02 kg m^{-3} from its value at 10 m. 199

200 2.3. Auxiliary datasets

201 We obtained 0.75°×0.75° gridded atmospheric/surface ocean reanalysis data at 3-hourly

- 202 temporal resolution from the ERA-Interim data product (Dee et al., 2011), which is produced
- by the European Centre for Medium-Range Weather Forecasts (ECMWF). Data were 203

- 204 downloaded from ECMWF on 1st August 2016 for the following variables, over the period
- 205 from 1st March 2014 to 31st December 2015: sea surface temperature (SST in °C), mean sea
- 206 level pressure (P_{baro} in bar), 10 metre U wind component (U_{10u} in m s⁻¹), 10 metre V wind
- 207 component (U_{10v} in m s⁻¹), and 2 metre dewpoint temperature (T_{dp} in °C). We calculated the
- wind speed at 10 m above the sea surface (U_{10} , in m s⁻¹) from its components U_{10u} and U_{10v} .
- 209 2.4. Central Celtic Sea site
- 210 Most of the analysis presented here was carried out at CCS, which was the most frequently-
- 211 occupied location during the UK-SSB sampling campaign. All samples collected within a
- 212 12 km radius of 49.4°N, 8.54°W were considered to represent CCS (Fig. 1). This radius was
- 213 selected such that sufficient data were captured to constrain the interior biogeochemical
- seasonality, whilst deviating as little as possible from the site itself. We then manually
- 215 grouped the CCS observational dataset based on the sampling date: we selected a set of date
- 216 boundaries that minimised the width of each time interval, while ensuring that there were
- 217 enough data within each interval to determine the vertical profile of each biogeochemical
- 218 variable, resulting in 15 time intervals (Table 1).
- Vertical profiles of biogeochemical variables at each site were typically homogeneous during the winter months, and had a two-layer shape during the stratified summer. As such, it was possible to fit error functions to many variables, in order to objectively determine values for each variable in the deep and surface layers, along with the depth of the divide. We thus generated a fit to all data for each variable at each time interval at each site using an equation of the form:

225
$$v = v_0 + v_1 \operatorname{erf}\left(\frac{z - z_0}{z_1}\right)$$
 (1)

- where *v* and *z* are the values and depths respectively of the measurements of each variable, erf is the Gauss error function, and v_0 , v_1 , z_0 and z_1 are coefficients that were adjusted to find the least-squares best fit to these data. These coefficients quantify useful properties: z_0 is the depth of the boundary between the surface and deep layers; z_1 indicates how sharp or diffuse the boundary is, with a smaller value indicating a transition over a narrower depth range; and the surface and deep layer values of the variable *v* are given by $(v_0 - v_1)$ and $(v_0 + v_1)$
- respectively (see Supp. Table S2 and Supp. Figs. S1–S4).

233 **2.5.** Air-sea CO₂ flux

- We calculated pCO_2^{sw} from the A_T , C_T , S, T, pressure, silicate and SRP observations using
- version 1.1 of the CO₂SYS program for MATLAB (van Heuven et al., 2011). We calculated
- Free scale pH in the same way (Supp. Fig. S5). We used the carbonic acid dissociation
- 237 constants of Lueker et al. (2000), the bisulfate dissociation constant of Dickson (1990) and
- the boron:chlorinity ratio of Lee et al. (2010). The uncertainties in the input variables resulted
- 239 in a 1σ precision of $\pm 8.1 \mu$ atm for calculated pCO_2^{sw} . Statistical interpolations of these
- 240 calculated pCO_2^{sw} values were limited by low temporal sampling resolution, resulting in key
- 241 features such as the 2014 spring bloom being partly missed, and therefore inaccurate
- estimates of the time-integrated air-sea CO₂ flux. We therefore created an empirical
- algorithm to predict the surface pCO_2^{sw} time series at CCS from the reanalysis SST data. The
- algorithm used different equations for different time periods and SST ranges (Table 2) to
- account for temporal variation in the surface pCO_2^{sw} -SST relationship resulting from changes
- in biological production and vertical mixing. The mean \pm SD difference between the
- 247 algorithm-predicted and measured pCO_2^{sw} was -4 ± 19 µatm across all 71 measurements
- above 25 m depth at CCS, which is similar to the uncertainties quoted for published pCO_2^{sw}
- algorithms in this region (e.g. Marrec et al., 2015).

We obtained measurements of the atmospheric CO₂ dry air mole fraction (*x*CO₂) for the
Greenhouse Gases Reference Network Site at Mace Head, County Galway, Ireland
(Dlugokencky et al., 2015). Geographically, this is the closest such observation site to our
study area. We applied a piecewise cubic Hermite interpolating polynomial (PCHIP) function
(Fritsch and Carlson, 1980; Kahaner et al., 1988) to their monthly mean data to predict *x*CO₂

- 255 in parts per million (ppm) for any given date. We estimated the uncertainty in xCO_2 values
- interpolated from the monthly means, ± 2.8 ppm, as the standard deviation of the residuals

257 between all of the original individual measurements and the interpolated xCO_2 values at the

- same time points. We then converted xCO_2 into atmospheric CO_2 partial pressure (pCO_2^{atm})
- using data from the ERA-Interim reanalysis (Dee et al., 2011) at CCS. First, we estimated the
- 260 water vapour pressure (P_v in bar) from T_{dp} (Alduchov and Eskridge, 1996; Lawrence, 2005):

261
$$P_{\rm v} = 610.94 \exp[17.625 T_{\rm dp} / (243.04 + T_{\rm dp})] 10^{-5}$$
 (2)

262 The atmospheric partial pressure of CO_2 (pCO_2^{atm} in µatm) was then given by:

263
$$pCO_2^{\text{atm}} = xCO_2 (P_{\text{baro}} - P_v) / 1.01325$$
 (3)

We calculated the air-sea CO₂ flux (*F*, in μ mol m⁻² hr⁻¹) following Wanninkhof (2014), using 3-hourly resolution data from ECMWF, the algorithm-predicted *p*CO₂^{sw} (Table 2), and *p*CO₂^{atm} calculated from *x*CO₂ (Eq. 3). The flux equation is:

267
$$F = k \alpha \left(p \operatorname{CO}_2^{\mathrm{sw}} - p \operatorname{CO}_2^{\mathrm{atm}} \right)$$
(4)

where α is the solubility of CO₂, evaluated following Weiss (1974) and converted to units of mol m⁻³ atm⁻¹ using the GSW toolbox (McDougall and Barker, 2011), and *k* is the gas transfer velocity in m hr⁻¹:

271
$$k = 0.00251 U_{10}^{2} (660/D)^{0.5}$$
 (5)

- 272 where D is the dimensionless Schmidt number for CO_2 at the surface layer seawater
- temperature (Wanninkhof, 2014). The flux calculations were carried out in MATLAB
- 274 (MathWorks) using the 'co2flux' program available from <u>https://github.com/mvdh7/co2flux</u>.

275 **2.6.** Column inventories

276 The column inventory of each biogeochemical variable was determined at each time interval by integrating the fitted profile for the variable (in mmol m^{-3}) from the surface to the seafloor 277 depth, taken as the mean seafloor depth across all CCS sampling stations (Z = 146.9 m). We 278 279 then attempted to remove the influence of processes other than NCP and remineralisation on 280 the column inventories. To correct for air-sea gas exchange, we subtracted the air-sea CO₂ flux integrated between each time point and winter 2014 (C_{atm}) from each time point's raw C_{T} 281 inventory. Hydes et al. (2004) suggest a constant denitrification rate of 0.1 mmol-N m⁻² day⁻¹ 282 for the Celtic Sea. We therefore subtracted the product of this rate and the number of days 283 284 between each time point and winter 2014 from the DIN inventory at each time point to correct for the potential loss of DIN through denitrification (DIN_{dnf}). Kitidis et al. (2017) 285 286 more recently reported a similar mean flux of DIN from sediments into the seawater due to 287 denitrification/anammox processes at a site near CCS. They identified some seasonal 288 variability in this value, but we did not include this effect as it was too small to alter our 289 analysis and conclusions.

290 **2.7.** Uncertainty analysis

291 The uncertainties in all of the final results, reported as 95% confidence intervals, were

- 292 calculated using Monte Carlo simulations. Explicitly, we replicated the column inventory
- analysis 1000 times, each time adding different random offsets to each individual

measurement. These offsets were normally distributed, with a SD equal to the 1σ analytical uncertainty for each measurement. The random noise was added both to the measurements of each variable. We report the 95% uncertainty in each column inventory as double the SD of all of its replicates.

298 2.8. Salinity correlations

299 We quantified the seasonally changing patterns in $C_{\rm T}$, DIN and SRP remineralisation across 300 the Celtic Sea by determining their correlations with practical salinity in the deep layer. For 301 these correlations, we used only data satisfying the following conditions: (1) salinity > 35.2, 302 to exclude near-coastal waters with nutrient-salinity relationships different from the majority 303 of the shelf; (2a) where seafloor depth < 200 m (i.e. on-shelf), sample depth > 60 m, to 304 exclude the surface layer where DIN and SRP are reduced to virtually zero in the summer 305 months; and (2b) where seafloor depth > 200 m (i.e. off-shelf), 150 m < sample depth < 200306 m, again to exclude the surface layer (which was deeper off-shelf) and also deeper waters that were less strongly affected by seasonal remineralisation. The potential density of the off-shelf 307 308 150 to 200 m layer throughout the year was similar to the on-shelf winter potential density, 309 supporting our choice of this as the off-shelf endmember for the correlations. Summary 310 statistics for these correlations are provided in Table 3, and scatter plots are show by Supp. 311 Figs. S12–S14.

313 **3. Results and discussion**

314 **3.1. Central Celtic Sea site**

315 *3.1.1. Air-sea CO*₂ *flux*

316 Few studies have examined the Celtic Sea inorganic carbon cycle in detail, but those that have were mostly focused on the surface layer and air-sea CO₂ exchange. Frankignoulle and 317 Borges (2001) used a compilation of surface ocean pCO_2^{sw} data to demonstrate that the 318 northwest European continental shelf seas are a net sink for atmospheric CO₂, at a rate 319 between 1.8 and 2.9 mol m^{-2} yr⁻¹. This result was quantitatively supported by a more recent 320 model study (Wakelin et al., 2012). However, other studies have found smaller net air-sea 321 322 CO_2 fluxes for this region. Borges et al. (2006) compiled published datasets across the entire northwest European shelf, and Marrec et al. (2015) generated predictive algorithms for Celtic 323 324 Sea pCO_2^{sw} from variables including sea surface temperature, mixed layer depth and 325 chlorophyll-a concentration. These studies agree that the Celtic Sea is a net sink for atmospheric CO₂, but they suggest a typical rate a little under 1 mol m^{-2} yr⁻¹. Our analysis 326 327 also indicated that the CCS site acted as a net atmospheric CO_2 sink: the mean air-to-sea CO_2 flux at CCS from 21st March 2014 (i.e. the start of the UK-SSB sampling period) to the same 328 date in 2015 was 1.3 ± 0.3 mol m⁻² yr⁻¹ (Fig. 2). The magnitude of this flux is consistent with 329 330 previous Celtic Sea studies, although it is towards their lower end. A separate analysis of in 331 situ CO₂ uptake (photosynthesis) and production (plankton respiration) data also found that 332 CCS was a net atmospheric CO_2 sink, as primary production was typically greater than

333 community respiration (García-Martín et al., 2017, this issue).

334 The $\Delta p CO_2$ was vertically uniform and close to atmospheric equilibrium during March-April.

335 Its seasonal pattern was similar to $C_{\rm T}$, with CO₂ undersaturation in the surface layer during

the spring-summer. The surface pCO_2^{sw} variability was dominated by biological C_T uptake

and release, rather than seasonal seawater heating and cooling, which would have been

338 expected to induce an increase in pCO_2^{sw} (and therefore ΔpCO_2) during the warmer summer

months. Indeed, the amplitude of the SST seasonal cycle was about 10°C. Alone, this would

- have driven a winter-summer pCO_2^{sw} increase of about 200 µatm (Takahashi et al., 2009),
- 341 whereas we observed a c. 85 µatm decline during this period (Fig. 2a). Other factors that
- 342 could affect the $\Delta p CO_2$ cycle are atmospheric $p CO_2$, and seawater A_T . The amplitude of
- 343 seasonal atmospheric pCO_2 cycle was an order of magnitude smaller than that of pCO_2^{sw} ,
- 344 while $A_{\rm T}$ was not noticeably affected by seasonal stratification, remaining relatively

- homogeneous both laterally and vertically throughout the Celtic Sea for the duration of the SSB sampling campaign (Supp. Fig. S8). The seasonal variability of atmospheric pCO_2 and A_T therefore did not significantly influence that of ΔpCO_2 .
- 348 The algorithm that we created to predict surface pCO_2^{sw} (Table 2) produced an annual range 349 of about 85 µatm, similar to previous studies (e.g. Marrec et al., 2015). However, there were differences in the seasonal pattern. Firstly, we found that the pCO_2^{sw} had two distinct minima 350 351 during 2014, first in May-June and then a weaker minimum in November-December associated with an autumn phytoplankton bloom (Wihsgott et al., 2017, this issue). Our 2015 352 353 data end too early in the seasonal cycle to determine whether these dual minima were 354 repeated. Secondly, Marrec et al. (2015) projected that $\Delta p CO_2$ should be positive for 355 approximately a third of each year, during the winter months, reaching values up to about 356 30 µatm. However, neither our algorithm nor our observations indicate such consistently high 357 $\Delta p CO_2$ values. We did not collect any carbonate chemistry samples between November and March, so we cannot be certain which is correct. However, the observational pCO_2^{sw} data 358 from SOCAT (Bakker et al., 2016) used by Marrec et al. (2015) to test their algorithm were 359 360 in atmospheric equilibrium during this season, rather than supersaturated as predicted, for 361 about half of the years that they considered. This winter supersaturation may therefore be an 362 intermittent phenomenon, but our dataset is unable to identify whether it occurred between 363 2014 and 2015.
- The most uncertain part of our air-sea CO_2 flux algorithm was between autumn 2014 and winter 2015, due to a lack of observations during that period. Comparison of the pattern of pCO_2^{sw} (Fig. 2a) with other studies (e.g. Marrec et al., 2015) and our understanding of marine carbonate system drivers suggests that the pCO_2^{sw} predicted by the algorithm for this period is likely to underestimate the true value. Consequently, we consider the air to sea CO_2 flux calculated for this period as a maximum estimate.

370 *3.1.2. Inorganic inventory changes*

Vertical profiles of $C_{\rm T}$, DIN, SRP and related hydrographic variables were constrained at two roughly equivalent time points both in 2014 and 2015, referred to as 'winter' (i.e. March-April, pre-spring bloom) and 'summer' (August). Between these, there was a single 'late spring' profile in June 2014, and a series of 'in-bloom' profiles (April-May) captured the progression of the 2015 spring bloom at a relatively high temporal resolution, followed by an 'early spring' profile in May 2015. There was an 'autumn' (November) profile in 2014 only. 377 Direct comparisons between the two years can therefore be drawn from comparing winter
378 and summer, while the additional points can be used to more finely resolve the seasonal
379 changes.

380 The TS chain measurements clearly illustrate the seasonal cycle of surface warming and 381 stratification followed by winter mixing (Fig. 3a). The CCS site was vertically mixed during 382 the winter, but physical and biogeochemical variables exhibited a two-layer stratified 383 structure during the spring and summer. Most variables were vertically homogeneous in 384 winter each year, after which $C_{\rm T}$, DIN and SRP declined strongly in the surface layer (i.e. 385 shallower than 40-50 m) through the spring and summer (Fig. 3b-d), through drawdown 386 associated with NCP. DIN reached virtually zero (i.e. beneath the detection limit of ~ 0.02 mmol m^{-3}), while SRP declined to a minimum of about 0.05 mmol m^{-3} , consistent with 387 previous studies that have identified DIN as the limiting nutrient in this region (e.g. Davis et 388 389 al., 2014). The concentrations of these variables increased in the deep layer over the same 390 period, consistent with remineralisation of sinking OM. Stratification began to break down 391 towards the autumn, and the water column was reset to a vertically homogeneous state by the 392 following winter.

393 Stratification had a less coherent effect on S (as measured by the TS chain), but here we saw 394 other forms of variability (Fig. 3e). In particular, there was a notable increase in deep layer S 395 (i.e. below about 50 m) from August to December 2014. This suggests that open ocean waters 396 increased in influence relative to on-shelf waters during this time, as no *in situ* processes can 397 modify deep layer S to this extent. The DIN and SRP increased disproportionately in the deep 398 layer during the same period as the aforementioned high S anomaly (Fig. 3c), suggesting that 399 increasing nutrient concentrations relative to $C_{\rm T}$ at this time may have been associated with 400 the advective flux that elevated S, rather than *in situ* OM remineralisation.

Deep layer salinity (i.e. beneath c. 50 m) increased from spring 2014 through to December
2014 (Fig. 3e) by an amount that cannot be accounted for by any *in situ* process. The DIN
and SRP also increased disproportionately in this deep layer during the same period (Fig. 3cd), suggesting that increasing nutrient concentrations at this time may have been associated
with an advective flux that elevated *S*, rather than *in situ* OM remineralisation. In support of
this, Ruiz et al. (2017, this issue) describe a shelf-wide estuarine-type circulation that advects
saline, nutrient-rich waters on to the shelf during the stratified summer months. They estimate

408 that 70-80% of the increase in deep layer DIN observed at CCS since the fully mixed winter409 state could be supplied by this physical transport mechanism.

410 3.1.3. Production and remineralisation

411 During the first half of each year, when stratification was increasing, changes in $C_{\rm T}$, DIN and 412 SRP profiles between successive time points can reveal the depth ranges over which NCP and 413 net remineralisation occurred (e.g. Fig. 4). For intervals when surface layer concentrations 414 declined, the average NCP rate can therefore be estimated by integrating the change in $C_{\rm T}$ 415 only over the depth range where $C_{\rm T}$ declined and correcting the result for air-sea CO₂ 416 exchange, which can be assumed to affect only the surface layer when the water column is 417 stratified. Similarly, deep layer increases in $C_{\rm T}$ can be attributed to remineralisation. The 418 same procedure can be carried out for DIN and SRP, with the remineralisation calculation for 419 DIN corrected for denitrification and/or anammox (Hydes et al., 2004; Kitidis et al., 2017), and no gas exchange correction (Table 4). The main spring bloom in 2014 resulted in a mean 420 NCP of 35.9 ± 2.3 mmol-C m⁻² d⁻¹ from winter to late spring, which is consistent with an 421 NCP estimate for the seasonally-stratified North Sea of 16-46 mmol $m^{-2} d^{-1}$ for this season 422 (Bozec et al., 2006). This represents a total of 3.08 ± 0.20 mol-C m⁻² C_T conversion into OM 423 424 since the mixed winter state, integrated over the period between our winter and late spring 425 2014 time points (i.e. 86 days; Table 1). The equivalent DIN and SRP drawdowns were $342 \pm$ 7 mmol $m^{-2} d^{-1}$ (Fig. 4a) and 20.74 ± 0.36 mmol $m^{-2} d^{-1}$ respectively, resulting in an overall 426 C:N:P for NCP of about 117:13.0:0.79. Corresponding remineralisation of $C_{\rm T}$, DIN and SRP 427 within the subsurface layer was 0.78 ± 0.50 mol m⁻², 85 ± 73 mmol m⁻² and 8 ± 99 mmol m⁻², 428 429 with C:N:P therefore 117:12.7:1.24. This indicates that 20-40% of the NCP was 430 remineralised within the sub-surface (presumably following sinking of the corresponding 431 OM), while 60–80% must have remained within some OM pool, assuming negligible net 432 advection (Pingree and Le Cann, 1989; Brown et al., 2003). The spring bloom period was followed by much weaker NCP and net water column remineralisation from late spring to 433 434 summer, and then vertical mixing into the autumn and winter (Fig. 4b-c), during which period the decline in deep layer $C_{\rm T}$ was due to mixing with low- $C_{\rm T}$ surface waters, not NCP. 435 436 The spring bloom of the following year was sampled at higher temporal resolution, during 437 cruise DY029 in April 2015. From winter to early spring 2015 (i.e. 61 days; Table 1), the

- 438 mean NCP determined from $C_{\rm T}$ profiles and air-sea CO₂ exchange was 30.5 ± 2.3 mmol-C
- 439 $m^{-2} d^{-1}$, comparable to the previous year, estimates from nearby similar regions (Bozec et al.,

2006), and the mean NCP of 31 ± 24 mmol-C m⁻² d⁻¹ (photosynthetic quotient of 1.4) 440 determined at CCS from oxygen-to-argon ratio (O₂/Ar) measurements during this cruise 441 442 (Seguro et al., 2017, this issue). This NCP integrated from the winter to early spring 2015 time points (i.e. 139 days; Table 1) represented 1.86 ± 0.14 mol-C m⁻² of OM production, 443 compared with 3.08 ± 0.20 mol-C m⁻² for the roughly equivalent period (winter to *late* 444 spring) in 2014. However, further NCP from early spring to summer in 2015 brought the total 445 NCP since winter to 3.30 ± 0.13 mol-C m⁻², similar to the total value for the part of 2014 with 446 positive NCP. The corresponding DIN- and SRP-derived values (from winter to summer 447 2015) were 343.4 \pm 3.9 and 20.86 \pm 0.22 mmol m⁻² respectively, resulting in a C:N:P for 448 449 NCP of about 117:12.2:0.74, similar to the previous year (i.e. 117:13.0:0.79), but with lower 450 N and P coefficients in 2014, or equivalently more C-rich OM production relative to typically 451 assumed (Redfield, 1934; Redfield et al., 1963) and measured (Anderson and Sarmiento, 452 1994) ratios. Therefore the total amount of OM generated through NCP appeared to be 453 similar during the spring bloom each year, although this may have occurred more rapidly 454 during 2014. Consequently, although the winter and summer time points were at a very 455 similar time for both years, it may be more appropriate to compare 'winter to late spring 456 2014' with 'winter to summer 2015'. Remineralisation from winter to summer 2015 occurred 457 at C:N:P of 117:11.7:0.88, and represented 24–35% of NCP, in both ways similar to 2014. 458 Both the production and remineralisation of OM were C-rich relative to the likely limiting 459 nutrient, with C:N ranging from 9 to 10 compared with typical values of 6.6 to 7.3 (Redfield, 460 1934; Redfield et al., 1963; Anderson and Sarmiento, 1994).

461 3.1.4. Total column inventories

462 The preceding section's approach of separating surface NCP from deep remineralisation was 463 less useful from summer onwards, once stratification had begun to break down. At this time, 464 both surface and deep layer $C_{\rm T}$ and nutrient inventories were influenced by vertical mixing, along with remineralisation throughout the water column. It is therefore more appropriate to 465 466 consider changes in the total inventories integrated vertically across the entire water column. Remineralisation of both $C_{\rm T}$ and nutrients continued after summer 2014, first with apparent 467 468 excess DIN generation until autumn 2014, and then excess $C_{\rm T}$ until winter 2015 (Fig. 5a-b). 469 We note that the excess N period coincides with the previously-noted apparent advective flux 470 of high salinity to CCS, high DIN waters in the deep layer, so this may have driven the 471 unexpected and transient high DIN, rather than remineralisation (Fig. 5c). However, this high 472 S water had disappeared by winter 2015, due to vertical mixing and shut-down of the summer 473 estuarine-type circulation (Ruiz-Castillo et al., 2017, this issue). The overall trajectory from

474 summer 2014 to winter 2015 is therefore probably representative of net remineralisation,475 presumably ongoing from some long-lived OM pool.

476 From late spring 2014 to winter 2015, remineralisation returned a further 2.17 ± 0.43 mol-C $m^{-2} C_T$, 169 ± 50 mmol-N m^{-2} DIN and 3.27 ± 0.88 mmol-P m^{-2} SRP to the water column, 477 478 corresponding to a C:N:P ratio of 117:9.1:0.2, with substantially lower N and P coefficients 479 and hence higher C:N and C:P than the NCP stoichiometry. Adding the additional apparent 480 late year remineralisation to that experienced in the deep layer from winter to late spring 481 2014 resulted in an overall remineralisation C:N:P of 117:10.0:0.5. This corresponded to 96 \pm 482 22 %, 72 ± 26 % and 56 ± 479 % of the OM production of $C_{\rm T}$, DIN and SRP respectively in the 2014 spring bloom being remineralised by winter 2015. In other words, almost all of the 483 484 $C_{\rm T}$ that was biologically fixed in the 2014 spring bloom was remineralised by winter 2015, 485 but only about three-quarters and half of the fixed DIN and SRP respectively were remineralised over the same period, while noting that the uncertainty in the SRP value is too 486 487 great for us to speculate about it further. As we previously described (Section 3.1.1), the air-488 sea CO₂ flux value that was used to correct these inventories was likely an overestimate from 489 autumn 2014 to winter 2015. Therefore the $C_{\rm T}$ result is a minimum estimate, and it is possible

490 that more $C_{\rm T}$ was remineralised than was produced by NCP that year at CCS.

491 These total inventories indicate that the shelf was not operating in steady state from winter 492 2014 to 2015. The winter inventories of DIN and SRP were 10.3 \pm 1.4 % and 10.6 \pm 1.1 % 493 lower respectively in 2015 compared to 2014, while that of $C_{\rm T}$ was 0.36 ± 0.18 % higher 494 (Fig. 5a). The equal (or slightly greater) OM formation during the 2015 spring-summer 495 relative to 2014 was therefore unexpected, because lower nutrient concentrations at the end of 496 the winter mixed state should have supported less NCP. As NCP was the ultimate driver of 497 CO₂ uptake, we would also have expected a lower air-to-sea CO₂ flux following these lower 498 winter nutrient concentrations. This did not occur, because the nutricline was about 10-15 m 499 deeper in 2015 (Fig. 6a), resulting in a greater total $N_{\rm T}$ conversion into OM despite its lower 500 initial concentration, and matching the behaviour of the thermocline (Fig. 6b). There does not 501 appear to be an obvious mechanism that would drive a deeper nutricline at times of lower 502 surface nutrient concentrations, which would lead summer NCP to be generally consistent 503 across different years. Rather, it appears to be a coincidence that these phenomena balanced

- 504 each other out for the two years that we observed.
 - 18

505 *3.1.5.* Organic matter stoichiometry

506 Overall mass balance dictates that changes in the OM composition should mirror the changes 507 in the inorganic inventories (corrected for gas exchange and denitrification/anammox). In 508 Fig. 5b, trajectories between adjacent sampling points that trend towards the bottom right 509 (low $C_{\rm T}$, high DIN or SRP) thus indicate the on-shelf generation of relatively C-rich OM, 510 while trajectories that tend towards the top left (high $C_{\rm T}$, low DIN or SRP) indicated more N-511 rich (and/or P-rich) OM generation, relative to an assumed 'typical' C:N:P of 117:16:1 512 (Anderson and Sarmiento, 1994). The high resolution samples from during the 2015 spring 513 bloom therefore suggest that NCP may have begun with N- and P-rich OM formation, before 514 switching to more C-rich production once surface layer DIN concentrations fell close to zero 515 (Fig. 5b). As mentioned above, the overall change from spring to summer 2015 was similar to 516 the C:N:P uptake stoichiometry observed for the same period in 2014. However, from early 517 spring to summer 2015 we saw the opposite pattern than during the similar period in 2014: 518 the OM became more C-rich. This change to seawater $C_{\rm T}$ loss with no nutrient loss (as 519 opposed to loss of both to OM following a constant stoichiometry) coincided approximately 520 with CCS surface DIN concentrations approaching zero. We suggest that the $C_{\rm T}$ loss was 521 therefore a result of continued NCP and OM formation, but that this was using regenerated 522 DIN within the surface layer as its N source rather than drawing from the inorganic pool, 523 which was by then locked away by stratification in the deep layer, as has been observed in the 524 North Sea (Bozec et al., 2006). Although this process should therefore have generated an 525 increasingly C-rich OM pool, presumably associated with more rapid N than C 526 remineralisation from the OM pool, it was not sufficient to return to the original winter 2014 527 C:N:P stoichiometry. These predicted changes in the OM composition are broadly in agreement with the pattern observed by direct OM stoichiometric measurements at CCS 528 529 (Davis et al., 2017, this issue), although our approach is not able to distinguish between the 530 dissolved and particulate OM pools.

531 **3.2.** Celtic Sea transect

532 *3.2.1.* Stratification across the Celtic Sea

533 In addition to the CCS site, we sampled a transect across the Celtic Sea during each UK-SSB

534 cruise. Analysis of these measurements adds a third dimension (i.e. distance from the open

- 535 ocean) to our so far two-dimensional analysis (i.e. depth and time), highlighting some
- 536 additional caveats, sources of uncertainty and broader context. Hartman et al. (2017, this

issue) describe surface layer UK-SSB measurements from across the entire northwestEuropean continental shelf, thus providing further context.

539 The pattern of winter mixing followed by surface layer uptake of $C_{\rm T}$, DIN and SRP and their 540 increase in the deep layer was consistent across the entire Celtic Sea transect (Fig. 7). 541 However, the deep layer experienced greater increases in these variables (and others such as 542 $\Delta p CO_2$) further away from the open ocean, although this was reduced for DIN by 543 denitrification/anammox in the seafloor sediments (Hydes et al., 2004; Kitidis et al., 2017). 544 The depth of the surface layer (as defined by the nutricline, 40-50 m) remained relatively 545 constant across the transect while the seafloor became shallower towards the land, so the 546 greater deep layer concentrations could simply indicate that the same amount of NCP had 547 been remineralised, but spread over a narrower deep layer, consistent with the limiting factor 548 for total remineralisation within the deep layer being some function of the OM supply. In 549 addition, the increasing deep layer $C_{\rm T}$ and nutrients away from the open ocean could reflect 550 the fact that deep waters further from the open ocean had experienced a longer transit time 551 across the shelf within the estuarine-type circulation (Ruiz-Castillo et al., 2017, this issue), 552 and thus greater accumulation of sinking organic matter from the surface layer, in a manner 553 analogous to how nutrient concentrations tend to increase along isopycnals away from the

- surface in the open ocean (Takahashi et al., 1985).
- 555 The increasing strength of biogeochemical stratification moving away from the open ocean 556 has important implications for interpreting our results from CCS, which lay towards the open
- 557 ocean end of the Celtic Sea transect. For example, the surface layer air-sea CO₂
- 558 disequilibrium during the stratified period increased away from the open ocean. We would
- therefore expect the Celtic Sea as a whole to exhibit a greater net air-sea CO₂ flux than that
- 560 calculated for the CCS site, which was relatively close to the open ocean end of the transect.
- 561 This is consistent with our CCS air-sea CO₂ flux estimate falling towards the lower end of
- 562 previously published results (Section 3.1.1). A second example concerns potential advective
- 563 fluxes within the Celtic Sea, as discussed in the following section.
- 564 3.2.2. Biogeochemical relationships with salinity

565 In the Celtic Sea, *S* acts as a tracer of the relative influence of open ocean and shelf waters,

- 566 particularly in the stratified deep layer where no other processes can modify it. Throughout
- 567 the year, *S* decreases moving on to the shelf, ultimately as a result of dilution by riverine
- 568 inputs. During the winter, *S* was vertically homogeneous at each point across the shelf, while

569 the isohalines were tilted away from vertical towards the open ocean in the surface during the 570 summer (i.e. increasing S with depth, Fig. 7h). This suggests that there was likely a net 571 transport towards the open ocean in the surface layer, and towards the land at depth at CCS 572 during the summer. During the winter, while $C_{\rm T}$ and SRP were laterally homogeneous, DIN 573 was correlated with salinity (Table 3) and declined away from the open ocean (Figs. 7d and 574 8b). There were no relationships between S and any of $C_{\rm T}$, DIN or SRP in the surface layer 575 during the summer months, when NCP was the dominant control on these biogeochemical 576 variables. However, correlations could be identified in the deep layer throughout the seasonal 577 cycle, which were variable through time, but fairly consistent with each other (Fig. 8, Table 578 3). These relationships with S indicated that $C_{\rm T}$ and SRP were enriched on the shelf relative 579 to the open ocean during the summer months, and returned close to open-ocean values during 580 the winter, consistent with a seasonal cycle of remineralisation and vertical mixing. Although 581 the DIN-S relationship had a similar pattern to those for $C_{\rm T}$ and SRP, there was a significant offset from the values predicted from the SRP relationship and an assumed N:P stoichiometry 582 of 16:1 (Fig. 8b). This offset was positive, with a mean value of $7.5 \pm 2.1 \text{ mmol m}^{-3}$ per unit 583 584 practical salinity, indicating that the shelf waters were virtually always depleted in DIN 585 relative to the open ocean, likely as the result of denitrification and/or anammox in the shelf 586 sediments (Hydes et al., 2004; Kitidis et al., 2017).

587 The preceding interpretation of the changing $C_{\rm T}$ and nutrient vertical profiles at CCS mostly ignored the possibility of advective influences. However, given the changing deep layer $C_{\rm T}$ 588 589 and nutrient concentrations in the stratified period, advection of water from another part of 590 the shelf could generate an apparent increase or decrease in remineralisation at CCS. As well 591 as directly changing the inorganic inventories, this could also supply OM to CCS that had 592 originally formed elsewhere on the shelf, particularly given our evidence that a significant 593 fraction of NCP (>50%) remained within the OM pool, as opposed to the deep layer 594 remineralised pool, over seasonal timescales (Fig. 5). Significant advection of a cross-shelf 595 OM gradient would hence introduce uncertainties into our one-dimensional estimates of the 596 fraction of NCP that had been remineralised. Indeed, production and advection of a long-597 lived OM pool could provide a source for the possible excess of remineralisation over NCP 598 that we observed from winter 2014 to winter 2015 at CCS. It is also important to note that we 599 cannot determine how CCS could have been affected by any advective fluxes of material 600 perpendicular to the Celtic Sea transect that we sampled, which could have a different 601 relationship with S than observed along the UK-SSB transect. However, in general we do not

- 602 expect this to have significantly affected our results, because previous studies have found that
- water exchange between the Celtic Sea and adjacent open ocean is relatively slow (Holt et al.,
- 604 2009; Huthnance et al., 2009), as is net lateral circulation on the shelf itself (Pingree and Le
- 605 Cann, 1989), with water taking about 2 years to reach the central Celtic Sea after entering the
- shelf from the open ocean (Hydes et al., 2004).
- 607 The Celtic Sea transect dataset also indicates that the increase in $C_{\rm T}$ observed at CCS from 608 winter 2014 to winter 2015, and the simultaneous declines in DIN and SRP, were not a result 609 of spatial variability. As these biogeochemical variables appeared to diverge from their open 610 ocean values moving into the shelf sea (Fig. 8), apparent changes at a static site such as CCS 611 could be caused by advective reorganisation of the water on the shelf, with a greater or lesser 612 open ocean influence at CCS in different years. If this were the case, then the apparent 613 nutrient loss that we observed could simply be an artefact of the spatial variability, rather than 614 a temporal biogeochemical trend. However, it appears that the changes in $C_{\rm T}$, DIN and SRP 615 that we observed at CCS over this time period were consistent throughout most of S range 616 observed across the Celtic Sea transect (Fig. 9). This therefore supports our earlier 617 interpretation that the seasonal biogeochemical changes observed at CCS were representative 618 of processes operating across the wider Celtic Sea.
- 619 The shelf $A_{\rm T}$ was relatively homogeneous both spatially and temporally throughout the UK-
- 620 SSB sampling campaign (Fig. 7i; Supp. Fig. S8), with a weakly positive relationship with
- 621 salinity. At CCS, we observed no low $A_{\rm T}$ excursions from the $A_{\rm T}$ -salinity relationship that
- 622 could be evidence for pelagic calcification, for example by coccolithophores (Suykens et al.,
- 623 2010). There was therefore no need to include a correction for this effect in calculating the
- 624 water column inventories of $C_{\rm T}$ at CCS through the UK-SSB sampling period.
- 625 3.2.3. Nutrient-conserving mechanism
- 626 Our estimated net oceanic CO_2 uptake of >1 mol-C m⁻² yr⁻¹ within the Celtic Sea suggests 627 that an active CSP operates within this system. Given the preceding presented data and
- 628 analysis, we suggest two mechanisms through which such a CSP could operate in a nutrient-
- 629 conserving manner, i.e. without a significant external nutrient-rich source from rivers, and
- 630 without the shelf becoming depleted in these nutrients over time, as follows.
- Firstly, we saw an increase in vertically-integrated $C_{\rm T}$ at CCS from winter 2014 to winter 2015, along with a simultaneous decline in DIN and SRP (Fig. 5a). The initial winter 2014

values of $C_{\rm T}$, DIN and SRP were relatively close to the base of the thermocline in the adjacent open ocean, suggesting that there may recently have been relatively strong water exchange across the shelf break, which did not occur again during our sampling period. If the shelf sea were to follow a cycle of increasing $C_{\rm T}$ while DIN and SRP declined between these irregular 'flushing' events, then the exchange of shelf waters with the open ocean would constitute a net export of $C_{\rm T}$ while DIN and SRP were replenished.

Secondly, as recently been identified in a similar environment in the nearby North Sea
(Johnson et al., 2013), it appears that a substantial fraction (>50%) of the OM produced
during the spring bloom each year was only slowly remineralised over many months (Fig. 5).
Indeed, evidence that deep layer advective transport could have accounted for 70-80% of the
DIN increase at CCS from winter to summer each year (Ruiz-Castillo et al., 2017, this issue)

- 644 suggests that our calculated fraction of OM that was not remineralised is a minimum
- 645 estimate, so the long-lived OM pool could be greater still. Moreover, the non-steady state

646 situation that we observed over the 1.5 sampled years further suggests that some fraction of

this (semi-)refractory OM pool was potentially not remineralised within the same year and

648 hence may play a role in redistributing C and nutrients within the Celtic Sea itself (Fig. 7).

- 649 We therefore speculate that this long-lived OM pool could also function to export C from the
- shelf sea into the open ocean, for example during exchange with off-shelf waters with a lower

651 OM concentration.

Future synthesis of the inorganic measurements presented here along with OM concentration

and stoichiometry datasets collected during the UK-SSB research programme in the context

of a system which likely operates on a multi-annual timescale will be required to assess the

655 capacity of these possible mechanisms to sustain a nutrient-conserving CSP in temperate,

656 seasonally stratified continental shelf seas like the Celtic Sea.

657 **4.** Conclusions

658 The Celtic Sea experiences seasonal physical and biogeochemical stratification with low $C_{\rm T}$ 659 and nutrient concentrations driving a net air to sea CO₂ flux, while remineralisation increases 660 these concentrations in the deep layer. Repeated observations at the CCS site suggest that the inventories of DIN and SRP declined by about 10% from winter 2014 to winter 2015, while 661 $C_{\rm T}$ increased. Transect data indicate that these temporal changes observed at CCS were 662 consistent across much of the Celtic Sea, rather than being an artefact of advection and spatial 663 664 variability. Lower availability of nutrients for NCP in 2015 was compensated by a deeper nutricline, so there was no reduction in either the total amount of NCP or air-sea CO₂ 665 exchange during the subsequent spring bloom. Analysis of stoichiometry during nutrient 666 uptake for NCP and remineralisation indicated the production of C-rich OM. However, some 667 668 fraction of the OM generated on the shelf appears to not have been remineralised within the 669 year it was created, suggested that a long-lived stock of OM could accumulate in the shelf 670 waters. This, and the overall changes in inorganic inventories that we observed, could provide 671 alternative mechanisms through which a nutrient-conserving CSP could operate, both in the 672 Celtic Sea and in other seasonally stratified, temperate continental shelf seas.

674 5. Acknowledgements

- 675 We thank the officers, crew and scientists involved with all UK-SSB cruises on the RRS
- 676 *Discovery* and RRS *James Cook*, in particular the carbonate chemistry samplers including C.
- 677 Kivimäe, L. Darroch, J. Fox, R. Houlding, P. Nelson, N. Hicks, A. Poulton, K. Mayers and R.
- 678 Sims. We thank I. Seguro, E. Ruiz-Castillo, E. García-Martín and C. Davis for useful
- 679 discussions. This study was funded by the Natural Environment Research Council (UK)
- 680 through "CaNDyFloSS: Carbon and Nutrient Dynamics and Fluxes over Shelf Systems"
- grants NE/K002058/1 to EMSW, NE/K001701/1 to JEH and SEH, and NE/K00185X/1 to
- 682 CMM and EPA.

684 6. References

- Alduchov, O.A., Eskridge, R.E., 1996. Improved Magnus Form Approximation of Saturation
 Vapor Pressure. J. Appl. Meteor. 35, 601–609. doi:10.1175/15200450(1996)035<0601:IMFAOS>2.0.CO;2
- Álvarez-Salgado, X.A., Doval, M.D., Borges, A.V., Frankignoulle, M., Woodward, E.M.S.,
 Figueiras, F.G., 2001. Off-shelf fluxes of labile materials by an upwelling filament in
 the NW Iberian Upwelling System. Prog. Oceanogr. 51, 321–337.
 doi:10.1016/S0079-6611(01)00073-8
- Anderson, L.A., Sarmiento, J.L., 1994. Redfield ratios of remineralization determined by
 nutrient data analysis. Global Biogeochem. Cy. 8, 65–80. doi:10.1029/93GB03318
- Bakker, D.C.E., Pfeil, B., Landa, C.S., Metzl, N., O'Brien, K.M., Olsen, A., Smith, K.,
 Cosca, C., Harasawa, S., Jones, S.D., Nakaoka, S.-I., Nojiri, Y., Schuster, U.,
 Steinhoff, T., Sweeney, C., Takahashi, T., Tilbrook, B., Wada, C., Wanninkhof, R.,
 Alin, S.R., Balestrini, C.F., Barbero, L., Bates, N.R., Bianchi, A.A., Bonou, F.,
- Boutin, J., Bozec, Y., Burger, E.F., Cai, W.-J., Castle, R.D., Chen, L., Chierici, M.,
- 699 Currie, K., Evans, W., Featherstone, C., Feely, R.A., Fransson, A., Goyet, C.,
- Greenwood, N., Gregor, L., Hankin, S., Hardman-Mountford, N.J., Harlay, J., Hauck,
 J., Hoppema, M., Humphreys, M.P., Hunt, C.W., Huss, B., Ibánhez, J.S.P.,
- 702 Johannessen, T., Keeling, R., Kitidis, V., Körtzinger, A., Kozyr, A., Krasakopoulou,
- E., Kuwata, A., Landschützer, P., Lauvset, S.K., Lefèvre, N., Lo Monaco, C., Manke,
- A., Mathis, J.T., Merlivat, L., Millero, F.J., Monteiro, P.M.S., Munro, D.R., Murata,
- A., Newberger, T., Omar, A.M., Ono, T., Paterson, K., Pearce, D., Pierrot, D.,
- 706 Robbins, L.L., Saito, S., Salisbury, J., Schlitzer, R., Schneider, B., Schweitzer, R.,
- 707 Sieger, R., Skjelvan, I., Sullivan, K.F., Sutherland, S.C., Sutton, A.J., Tadokoro, K.,
- 708 Telszewski, M., Tuma, M., van Heuven, S.M.A.C., Vandemark, D., Ward, B.,
- Watson, A.J., Xu, S., 2016. A multi-decade record of high-quality *f*CO₂ data in
 version 3 of the Surface Ocean CO₂ Atlas (SOCAT). Earth Syst. Sci. Data 8, 383–
- 711 413. doi:10.5194/essd-8-383-2016
- Borges, A.V., 2005. Do we have enough pieces of the jigsaw to integrate CO₂ fluxes in the coastal ocean? Estuaries 28, 3–27. doi:10.1007/BF02732750Borges, A.V.,
 Schiettecatte, L.-S., Abril, G., Delille, B., Gazeau, F., 2006. Carbon dioxide in European coastal waters. Estuar. Coast. Shelf S. 70, 375–387.
- 716 doi:10.1016/j.ecss.2006.05.046
- Bozec, Y., Thomas, H., Schiettecatte, L.-S., Borges, A.V., Elkalay, K., de Baar, H.J.W.,
 2006. Assessment of the processes controlling the seasonal variations of dissolved
 inorganic carbon in the North Sea. Limnol. Oceanogr. 51, 2746–2762.
 doi:10.4319/lo.2006.51.6.2746
- Broecker, W.S., Peng, T.-H., 1974. Gas exchange rates between air and sea. Tellus 26, 21–
 35. doi:10.3402/tellusa.v26i1-2.9733Brown, J., Carrillo, L., Fernand, L., Horsburgh,
 K.J., Hill, A.E., Young, E.F., Medler, K.J., 2003. Observations of the physical
 structure and seasonal jet-like circulation of the Celtic Sea and St. George's Channel
 of the Irish Sea. Cont. Shelf Res. 23, 533–561. doi:10.1016/S0278-4343(03)00008-6
- Chen, C.-T.A., Borges, A.V., 2009. Reconciling opposing views on carbon cycling in the
 coastal ocean: Continental shelves as sinks and near-shore ecosystems as sources of
 atmospheric CO₂. Deep-Sea Res. Pt II 56, 578–590. doi:10.1016/j.dsr2.2009.01.001
- Chen, C.-T.A., Huang, T.-H., Chen, Y.-C., Bai, Y., He, X., Kang, Y., 2013. Air–sea
 exchanges of CO₂ in the world's coastal seas. Biogeosciences 10, 6509–6544.
 doi:10.5194/bg-10-6509-2013

- Cooper, L.H.N., Vaux, D., 1949. Cascading Over the Continental Slope of Water from the
 Celtic Sea. J. Mar. Biol. Ass. U.K. 28, 719–750. doi:10.1017/S0025315400023547
- Dall'Olmo, G., Dingle, J., Polimene, L., Brewin, R.J.W., Claustré, H., 2016. Substantial
 energy input to the mesopelagic ecosystem from the seasonal mixed-layer pump.
 Nature Geosci. 9, 820. doi:10.1038/ngeo2818
- Davis, C.E., Blackbird, S., Wolff, G., Woodward, M., Mahaffey, C., 2017. Seasonal organic
 matter dynamics in a temperate shelf sea. Prog. Oceanogr. in review.
- Davis, C.E., Mahaffey, C., Wolff, G.A., Sharples, J., 2014. A storm in a shelf sea: Variation
 in phosphorus distribution and organic matter stoichiometry. Geophys. Res. Lett. 41,
 2014GL061949. doi:10.1002/2014GL061949
- de Haas, H., van Weering, T.C.E., de Stigter, H., 2002. Organic carbon in shelf seas: sinks or
 sources, processes and products. Cont. Shelf Res. 22, 691–717. doi:10.1016/S02784343(01)00093-0
- Dee, D.P., Uppala, S.M., Simmons, A.J., Berrisford, P., Poli, P., Kobayashi, S., Andrae, U.,
 Balmaseda, M.A., Balsamo, G., Bauer, P., Bechtold, P., Beljaars, A.C.M., van de
 Berg, L., Bidlot, J., Bormann, N., Delsol, C., Dragani, R., Fuentes, M., Geer, A.J.,
 Haimberger, L., Healy, S.B., Hersbach, H., Hólm, E.V., Isaksen, L., Kållberg, P.,
 Köhler, M., Matricardi, M., McNally, A.P., Monge-Sanz, B.M., Morcrette, J.-J., Park,
- 749 Koller, M., Walicardi, M., WeiNary, A.F., Molige-Salz, B.M., Molerette, J.-J., Park 750 B.-K., Peubey, C., de Rosnay, P., Tavolato, C., Thépaut, J.-N., Vitart, F., 2011. The
- ERA-Interim reanalysis: configuration and performance of the data assimilation
 system. Q. J. R. Meteorol. Soc. 137, 553–597. doi:10.1002/qj.828
- 753Dickson, A.G., 1990. Standard potential of the reaction: $AgCl_{(s)} + 0.5 H_{2(g)} = Ag_{(s)} + HCl_{(aq)}$,754and the standard acidity constant of the ion HSO_4^- in synthetic sea water from 273.15755to 318.15 K. J. Chem. Thermodyn. 22, 113–127. doi:10.1016/0021-9614(90)90074-Z
- Dickson, A.G., Afghan, J.D., Anderson, G.C., 2003. Reference materials for oceanic CO₂
 analysis: a method for the certification of total alkalinity. Mar. Chem. 80, 185–197.
 doi:10.1016/S0304-4203(02)00133-0
- Dickson, A.G., Sabine, C.L., Christian, J.R. (Eds.), 2007. Guide to best practices for ocean
 CO₂ measurements. PICES Special Publication 3.
- Diesing, M., Kröger, S., Parker, R., Jenkins, C., Mason, C., Weston, K., 2017. Predicting the
 standing stock of organic carbon in surface sediments of the North–West European
 continental shelf. Biogeochemistry 1–18. doi:10.1007/s10533-017-0310-4
- Dlugokencky, E.J., Lang, P.M., Masarie, K.A., Crotwell, A.M., Crotwell, M.J., 2015.
 Atmospheric Carbon Dioxide Dry Air Mole Fractions from the NOAA ESRL Carbon
 Cycle Cooperative Global Air Sampling Network, 1968-2014, Version: 2015-08-03.
 ftp://aftp.cmdl.noaa.gov/data/trace_gases/co2/flask/surface/.
- Frankignoulle, M., Borges, A.V., 2001. European continental shelf as a significant sink for
 atmospheric carbon dioxide. Global Biogeochem. Cy. 15, 569–576.
 doi:10.1029/2000GB001307
- Fritsch, F., Carlson, R., 1980. Monotone piecewise cubic interpolation. SIAM J. Numer.
 Anal. 17, 238–246. doi:10.1137/0717021García-Martín, E.E., Daniels, C.J.,
 Davidson, K., Lozano, J., Mayers, K.M.J., McNeill, S., Mitchell, E., Poulton, A.J.,
- Purdie, D.A., Tarran, G.A., Whyte, C., Robinson, C., 2017. Plankton community
 respiration and bacterial metabolism in a North Atlantic Shelf Sea during spring
 bloom development (April 2015). Prog. Oceanogr. doi:10.1016/j.pocean.2017.11.002
- Gowen, R.J., Hydes, D.J., Mills, D.K., Stewart, B.M., Brown, J., Gibson, C.E., Shammon,
 T.M., Allen, M., Malcolm, S.J., 2002. Assessing Trends in Nutrient Concentrations in
 Coastal Shelf Seas: a Case Study in the Irish Sea. Estuar. Coast. Shelf S. 54, 927–939.
 doi:10.1006/ecss.2001.0849

781 Hartman, S., Humphreys, M., Kivimäe, C., Woodward, M., Kitidis, V., McGrath, T., Hydes, 782 D., Greenwood, N., Ostle, C., Pearce, D., Sivyer, D., Stewart, B., Walsham, P., McGovern, E., Harris, C., Griffiths, A., Smilenova, A., Clarke, J., Davis, C., 783 784 Nightingale, P., 2017. Seasonality and spatial heterogeneity of the surface water

785 carbonate system on the NW European shelf. Prog. Oceanogr. in press.

- Humphreys, M.P., 2015a. Calculating seawater total alkalinity from open-cell titration data 786 787 using a modified Gran plot technique, in: Measurements and Concepts in Marine 788 Carbonate Chemistry. PhD Thesis, Ocean and Earth Science, University of 789 Southampton, UK, pp. 25-44.
- 790 Humphreys, M.P., 2015b. Cross-over analysis of hydrographic variables: XOVER v1.0. 791 Ocean and Earth Science, University of Southampton, UK pp 8. 792 doi:10.13140/RG.2.1.1629.0405
- 793 Hydes, D.J., Gowen, R.J., Holliday, N.P., Shammon, T., Mills, D., 2004. External and 794 internal control of winter concentrations of nutrients (N, P and Si) in north-west 795 European shelf seas. Estuar. Coast. Shelf S. 59, 151–161.

796 doi:10.1016/j.ecss.2003.08.004

- 797 IPCC, 2013. Climate Change 2013: The Physical Science Basis. Contribution of Working 798 Group I to the Fifth Assessment Report of the Intergovernmental Panel on Climate 799 Change. Cambridge University Press, Cambridge, United Kingdom and New York, 800 NY, USA.
- 801 Ivanov, V.V., Shapiro, G.I., Huthnance, J.M., Aleynik, D.L., Golovin, P.N., 2004. Cascades 802 of dense water around the world ocean. Prog. Oceanogr. 60, 47-98. 803 doi:10.1016/j.pocean.2003.12.002
- 804 Johnson, M.T., Greenwood, N., Sivyer, D.B., Thomson, M., Reeve, A., Weston, K., Jickells, 805 T.D., 2013. Characterising the seasonal cycle of dissolved organic nitrogen using 806 Cefas SmartBuoy high-resolution time-series samples from the southern North Sea. Biogeochemistry 113, 23-36. doi:10.1007/s10533-012-9738-8 807
- 808 Jones, D.C., Ito, T., Takano, Y., Hsu, W.-C., 2014. Spatial and seasonal variability of the air-809 sea equilibration timescale of carbon dioxide. Global Biogeochem. Cy. 28, 1163-810 1178. doi:10.1002/2014GB004813
- 811 Kahaner, D., Moler, C., Nash, S., 1988. Numerical Methods and Software. Prentice Hall, 812 Englewood Cliffs, NJ, U.S.A.
- 813 Kitidis, V., Tait, K., Nunes, J., Brown, I., Woodward, E.M.S., Harris, C., Sabadel, A.J.M., 814 Sivyer, D.B., Silburn, B., Kröger, S., 2017. Seasonal benthic nitrogen cycling in a 815 temperate shelf sea: the Celtic Sea. Biogeochemistry 1-17. doi:10.1007/s10533-017-816 0311-3
- Laruelle, G.G., Lauerwald, R., Pfeil, B., Regnier, P., 2014. Regionalized global budget of the 817 818 CO₂ exchange at the air-water interface in continental shelf seas. Global Biogeochem. 819 Cy. 28, 2014GB004832. doi:10.1002/2014GB004832
- 820 Lawrence, M.G., 2005. The Relationship between Relative Humidity and the Dewpoint 821 Temperature in Moist Air: A Simple Conversion and Applications. Bull. Amer. 822 Meteor. Soc. 86, 225-233. doi:10.1175/BAMS-86-2-225
- 823 Le Menn, M., 2011. About uncertainties in practical salinity calculations. Ocean Sci. 7, 651-824 659. doi:10.5194/os-7-651-2011
- 825 Le Quéré, C., Raupach, M.R., Canadell, J.G., Marland, G., Bopp, L., Ciais, P., Conway, T.J., 826 Doney, S.C., Feely, R.A., Foster, P., Friedlingstein, P., Gurney, K., Houghton, R.A., 827 House, J.I., Huntingford, C., Levy, P.E., Lomas, M.R., Majkut, J., Metzl, N., Ometto,
- 828 J.P., Peters, G.P., Prentice, I.C., Randerson, J.T., Running, S.W., Sarmiento, J.L.,
- 829 Schuster, U., Sitch, S., Takahashi, T., Viovy, N., van der Werf, G.R., Woodward, F.I.,

- 830 2009. Trends in the sources and sinks of carbon dioxide. Nature Geosci. 2, 831–836.
 831 doi:10.1038/ngeo689
- Lee, K., Kim, T.-W., Byrne, R.H., Millero, F.J., Feely, R.A., Liu, Y.-M., 2010. The universal ratio of boron to chlorinity for the North Pacific and North Atlantic oceans. Geochim.
 Cosmochim. Acta 74, 1801–1811. doi:10.1016/j.gca.2009.12.027
- Lueker, T.J., Dickson, A.G., Keeling, C.D., 2000. Ocean *p*CO₂ calculated from dissolved
 inorganic carbon, alkalinity, and equations for K₁ and K₂: validation based on
 laboratory measurements of CO₂ in gas and seawater at equilibrium. Mar. Chem. 70,
 105–119. doi:10.1016/S0304-4203(00)00022-0
- Marrec, P., Cariou, T., Macé, E., Morin, P., Salt, L.A., Vernet, M., Taylor, B., Paxman, K.,
 Bozec, Y., 2015. Dynamics of air–sea CO₂ fluxes in the northwestern European shelf
 based on voluntary observing ship and satellite observations. Biogeosciences 12,
 5371–5391. doi:10.5194/bg-12-5371-2015
- McDougall, T.J., Barker, P.M., 2011. Getting started with TEOS-10 and the Gibbs Seawater
 (GSW) Oceanographic Toolbox. SCOR/IAPSO WG127.
- McGrath, T., McGovern, E., Cave, R.R., Kivimäe, C., 2015. The Inorganic Carbon
 Chemistry in Coastal and Shelf Waters Around Ireland. Estuar. Coast. 1–13.
 doi:10.1007/s12237-015-9950-6
- Olsen, A., Key, R.M., Heuven, S. van, Lauvset, S.K., Velo, A., Lin, X., Schirnick, C., Kozyr,
 A., Tanhua, T., Hoppema, M., Jutterström, S., Steinfeldt, R., Jeansson, E., Ishii, M.,
 Pérez, F.F., Suzuki, T., 2016. The Global Ocean Data Analysis Project version 2
 (GLODAPv2) an internally consistent data product for the world ocean. Earth Syst.
 Sci. Data 8, 297–323. doi:10.5194/essd-8-297-2016
- Pingree, R.D., Le Cann, B., 1989. Celtic and Armorican slope and shelf residual currents.
 Prog. Oceanogr. 23, 303–338. doi:10.1016/0079-6611(89)90003-7
- Redfield, A.C., 1934. On the proportions of organic derivatives in sea water and their relation
 to the composition of plankton, in: James Johnstone Memorial Volume. University
 Press of Liverpool, pp. 176–192.
- Redfield, A.C., Ketchum, B.H., Richards, F.A., 1963. The influence of organisms on the
 composition of sea-water, in: The Sea. Interscience, New York, pp. 26–77.
- Ruiz-Castillo, E., Sharples, J., Hopkins, J.E., Woodward, E.M.S., 2017. Seasonality in the
 cross-shelf physical structure and the implications for nutrients supplies to a
 temperate shelf sea. Prog. Oceanogr. in review.
- Seguro, I., Marca, A.D., Painting, S.J., Shutler, J.D., Suggett, D.J., Kaiser, J., 2017. Highresolution net and gross biological production during a Celtic Sea spring bloom. Prog.
 Oceanogr. doi:10.1016/j.pocean.2017.12.003
- Shapiro, G.I., Huthnance, J.M., Ivanov, V.V., 2003. Dense water cascading off the
 continental shelf. J. Geophys. Res. 108, 3390. doi:10.1029/2002JC001610
- Suykens, K., Delille, B., Chou, L., De Bodt, C., Harlay, J., Borges, A.V., 2010. Dissolved
 inorganic carbon dynamics and air-sea carbon dioxide fluxes during coccolithophore
 blooms in the northwest European continental margin (northern Bay of Biscay).
 Global Biogeochem. Cy. 24, GB3022. doi:10.1029/2009GB003730
- Suykens, K., Schmidt, S., Delille, B., Harlay, J., Chou, L., De Bodt, C., Fagel, N., Borges,
 A.V., 2011. Benthic remineralization in the northwest European continental margin
 (northern Bay of Biscay). Cont. Shelf Res. 31, 644–658.
- doi:10.1016/j.csr.2010.12.017Takahashi, T., Broecker, W.S., Langer, S., 1985.
- Redfield Ratio Based on Chemical Data from Isopycnal Surfaces. J. Geophys. Res.
 90, 6907–6924. doi:198510.1029/JC090iC04p06907
- Takahashi, T., Sutherland, S.C., Wanninkhof, R., Sweeney, C., Feely, R.A., Chipman, D.W.,
 Hales, B., Friederich, G., Chavez, F., Sabine, C., Watson, A., Bakker, D.C.E.,

880	Schuster, U., Metzl, N., Yoshikawa-Inoue, H., Ishii, M., Midorikawa, T., Nojiri, Y.,
881	Körtzinger, A., Steinhoff, T., Hoppema, M., Olafsson, J., Arnarson, T.S., Tilbrook,
882	B., Johannessen, T., Olsen, A., Bellerby, R., Wong, C.S., Delille, B., Bates, N.R., de
883	Baar, H.J.W., 2009. Climatological mean and decadal change in surface ocean pCO_2 ,
884	and net sea-air CO_2 flux over the global oceans. Deep-Sea Res. Pt II 56, 554–577.
885	doi:10.1016/j.dsr2.2008.12.009
886	Thomas, H., Bozec, Y., Elkalay, K., Baar, H.J.W. de, 2004. Enhanced Open Ocean Storage
887	of CO ₂ from Shelf Sea Pumping. Science 304, 1005–1008.
888	doi:10.1126/science.1095491
889	Tsunogai, S., Watanabe, S., Sato, T., 1999. Is there a "continental shelf pump" for the
890	absorption of atmospheric CO ₂ ? Tellus B 51, 701–712. doi:10.1034/j.1600-
891	0889.1999.t01-2-00010.x
892	van Heuven, S., Pierrot, D., Rae, J.W.B., Lewis, E., Wallace, D.W.R., 2011. CO ₂ SYS v 1.1,
893	MATLAB program developed for CO ₂ system calculations. ORNL/CDIAC-105b.
894	Carbon Dioxide Information Analysis Center, Oak Ridge National Laboratory, U.S.
895	Department of Energy, Oak Ridge, TN, USA.
896	Volk, T., Hoffert, M.I., 1985. Ocean Carbon Pumps: Analysis of Relative Strengths and
897	Efficiencies in Ocean-Driven Atmospheric CO ₂ Changes, in: Sundquist, E.T.,
898	Broecker, W.S. (Eds.), The Carbon Cycle and Atmospheric CO ₂ : Natural Variations
899	Archean to Present. American Geophysical Union, USA, pp. 99–110.
900	Wakelin, S.L., Holt, J.T., Blackford, J.C., Allen, J.I., Butenschön, M., Artioli, Y., 2012.
901	Modeling the carbon fluxes of the northwest European continental shelf: Validation
902	and budgets. J. Geophys. Res. Oceans 117, C05020. doi:10.1029/2011JC007402
903	Wanninkhof, R., 2014. Relationship between wind speed and gas exchange over the ocean
904	revisited: Gas exchange and wind speed over the ocean. Limnol. Oceanogr. Methods
905	12, 351–362. doi:10.4319/lom.2014.12.351
906	Weiss, R.F., 1974. Carbon dioxide in water and seawater: the solubility of a non-ideal gas.
907	Mar. Chem. 2, 203–215. doi:10.1016/0304-4203(74)90015-2
908	Wihsgott, J., Hopkins, J., Sharples, J., Jones, E., Balfour, C.A., 2016. Long-term mooring
909	observations of full depth water column structure spanning 17 months, collected in a
910	temperate shelf sea (Celtic Sea). British Oceanographic Data Centre, Natural
911	Environment Research Council, UK. doi:10/bqwf
912	Wihsgott, J.U., Sharples, J., Hopkins, J.E., Woodward, E.M.S., Greenwood, N., Hull, T.,
913	Sivyer, D.B., 2017. Investigating the autumn bloom's significance within the seasonal
914	cycle of primary production in a temperate shelf sea. Prog. Oceanogr. in review.
915	Wollast, R., 1998. Evaluation and comparison of the global carbon cycle in the coastal zone
916	and in the open ocean, in: Brink, K. H., Robinson, A. R. (Eds.), The Global Coastal
917	Ocean. John Wiley and Sons, New York, pp. 213–252.
918	Woodward, E.M.S., Rees, A.P., 2001. Nutrient distributions in an anticyclonic eddy in the
919	northeast Atlantic Ocean, with reference to nanomolar ammonium concentrations.
920	Deep-Sea Res. Pt II 48, 775–793. doi:10.1016/S0967-0645(00)00097-7
921	

7. Tables

Name of time	Year	Interval	Sampling date range	Cruise(s)
interval		code	(dd/mm/yyyy)	
Winter	2014	W14	26/03/2014 - 27/03/2014	DY008
Late Spring	2014	lSp14	19/06/2014 - 22/06/2014	JC105
Summer	2014	Su14	05/08/2014 - 23/08/2014	DY026A,
				DY026B
Autumn	2014	A14	10/11/2014 - 29/11/2014	DY018
Winter	2015	W14	22/03/2015	DY021
In-bloom 1	2015	IB1	03/04/2015 - 06/04/2015	DY029
In-bloom 2	2015	IB2	11/04/2015 - 12/04/2015	DY029
In-bloom 3	2015	IB3	15/04/2015	DY029
In-bloom 4	2015	IB4	16/04/2015	DY029
In-bloom 5	2015	IB5	20/04/2015	DY029
In-bloom 6	2015	IB6	21/04/2015	DY029
In-bloom 7	2015	IB7	25/04/2015 - 26/04/2015	DY029
In-bloom 8	2015	IB8	28/04/2015	DY029
Early Spring	2015	eSp15	21/05/2015 - 23/05/2015	DY030
Summer	2015	Su15	24/07/2015 - 23/08/2015	DY033, DY034

Table 1. Summary of the time intervals used in the CCS site analysis.

927 Table 2. Details of the empirical algorithm used to predict surface pCO_2^{sw} at CCS.

Day of year (DoY) range	SST range / °C	$p \text{CO}_2^{\text{sw}} / \mu \text{atm}$
80 < DoY < 232	SST < 10.5	$1.547 \times 10^{-3} \exp(\text{SST}) + 372.7$
80 < DoY < 232	$10.5 \le \text{SST} \le 11.0$	$324.5 + 6.596 \times 10^{24} \exp[-5.000(\text{SST})]$
80 < DoY < 232	11.0 < SST	$-0.557(SST)^2 + 23.57(SST) + 141.1$
$DoY \le 80 \text{ or } 233 \le DoY$	All	$-0.2479(SST)^3 + 10.846(SST)^2$
		- 151.42(SST) + 1050.3

931

Table 3. Correlation statistics between biogeochemical variables and salinity. Columns headed dX/dS contain gradient ± its standard error in µmol kg⁻¹ (Figure 8); *N* is the number of observations used in each regression.

Cruise	Interval code	C_{T}				DIN			SRP				
		$\mathrm{d}C_{\mathrm{T}}/\mathrm{d}S$	r^2	р	N	dDIN/dS	r^2	р	N	dSRP/dS	r^2	р	N
DY008	W14	-10.6 ± 8.9	0.11	0.26	13	7.6 ± 0.6	0.86	0.00	25	-0.23 ± 0.09	0.21	0.02	25
JC105	lSp14	-109.6 ± 10.0	0.86	0.00	21	1.0 ± 0.8	0.07	0.24	21	-0.46 ± 0.06	0.75	0.00	21
DY026*	Su14	-60.6 ± 10.8	0.78	0.00	11	4.1 ± 1.2	0.36	0.00	23	-0.29 ± 0.06	0.47	0.00	26
DY018	A14	-99.5 ± 10.6	0.49	0.00	92	-0.2 ± 1.1	0.00	0.89	183	-0.46 ± 0.07	0.18	0.00	202
DY021	W15	-19.6 ± 9.1	0.28	0.05	14	5.9 ± 0.7	0.59	0.00	47	0.01 ± 0.06	0.00	0.87	47
DY029	IB1–IB8	-39.5 ± 7.3	0.54	0.00	27	7.4 ± 0.4	0.65	0.00	192	0.11 ± 0.02	0.12	0.00	202
DY030	eS15	-26.7 ± 13.4	0.50	0.12	6	9.5 ± 1.2	0.85	0.00	13	0.32 ± 0.10	0.48	0.01	13
DY033	Su15	-63.1 ± 8.1	0.68	0.00	31	4.2 ± 0.4	0.42	0.00	182	-0.21 ± 0.02	0.30	0.00	186
DY034	Su15	-120.9 ± 24.0	0.78	0.00	9	0.4 ± 1.8	0.00	0.84	44	-0.51 ± 0.12	0.31	0.00	44

934 *DY026 includes all data from both DY026A and DY026B.

936 Table 4.

Start	End	Туре	С	Ν	Р
W14	lSp14	NCP			
W15	eSp15	NCP			
W15	Su15	NCP			





Figure 1. The UK-SSB sampling locations in the Celtic Sea. The pink circle encloses the data considered to represent the Central Celtic Sea site (i.e. within 12 km of 49.4°N, 8.54°W), while the yellow lines indicate the transect route for Fig. 7 and enclose the data included in those plots; the points labelled A, B and C correspond to the equivalent points on Fig. 7. The 200 m bathymetry contour, traditionally used to define the edge of the shelf, is shown as a black line. Topography data are from the GEBCO_2014 30 arc-second grid (version 20150318, <u>http://www.gebco.net</u>).





Figure 2. Time-series of air-sea CO_2 disequilibrium and flux at CCS. (a) Atmospheric (red) and sea surface (blue) pCO_2 at CCS. The black points show the actual measurements of

951 surface pCO_2^{sw} , while the blue line shows our algorithm-predicted interpolation of these data

952 (Table 2) that was used to calculate $\Delta p CO_2$ and then the air-sea CO_2 flux. (b) The sea to air

953 CO_2 flux. The shaded areas show the 95% confidence intervals.



967 Figure 3. Hovmöller diagrams of CCS water column (a) temperature, (b) dissolved inorganic 968 carbon, (c) dissolved inorganic nitrogen (i.e. nitrate + nitrate + ammonium), (d) soluble 969 reactive phosphorus, (e) practical salinity, and (f) air-sea CO₂ disequilibrium. Where shown, 970 sampling locations are indicated as black points. The cycle of winter mixing and summer 971 stratification is clear for most variables, while *S*, DIN and SRP show an anomalous increase 972 in the deep layer during late 2014, indicating an influx of open ocean waters.



977 Figure 4. Changes in the DIN profile at CCS for selected time intervals. In each panel, the 978 earlier profile is shown as a dashed line, while the later one is solid. The blue shaded areas 979 therefore indicate decreases in DIN, while orange shows increases. (a) Primary production in 980 the surface layer drew down DIN, while remineralisation at depth increased the DIN 981 concentration. (b) The onset of vertical mixing eroded the surface-deep DIN gradient, 982 overlaid by a remineralisation-driven increase in the total DIN inventory. (c) The water 983 column returned to a vertically homogeneous state following further vertical mixing.

985





999 Figure 5. Water column inventories at CCS throughout the SSB study period. The inventories 1000 have been divided by the water depth, so the values shown are the vertical mean at each time 1001 point. Downwards/upwards pointing triangles represent 2014/2015 data respectively, and the 1002 grey diagonal lines indicate a C:N:P stoichiometry of 117:16:1 (Anderson and Sarmiento, 1003 1994). W = winter; eSp = early spring; lSp = late spring; Su = summer; A = autumn; 14/15 1004 indicates the year. (a) Raw inventories of $C_{\rm T}$ and $N_{\rm T}$ measured at CCS, coloured by the 1005 sampling date. (b) Here, the $C_{\rm T}$ inventory is corrected for air-sea gas exchange relative to the 1006 first sampling point, while the DINis corrected for denitrification/anammox. The colour 1007 shows the surface layer DIN concentration. (c) Relative changes in DIN (corrected for 1008 denitrification/anammox) and SRP, coloured by mean salinity at CCS at each time point. 1009 Higher salinity values indicate the influence of open ocean waters.



Figure 6. Winter to summer changes in (a) DIN and (b) seawater temperature for 2014 (blue) and 2015 (red). In (a), the dashed lines show the lowest possible Δ DIN, i.e. its value if the entire winter DIN inventory had been converted into OM. All available DIN in the surface layer was converted to OM in both years. This represented a greater change in DIN in 2014 due to the higher initial DIN concentration, but the deeper nutricline in 2015 led to greater NCP overall. Remineralisation increased the deep layer DIN concentration by the same amount in both years, but this represented a smaller absolute DIN addition in 2015 owing to the deeper nutricline.



1037 Figure 7. Transects across the Celtic Sea for selected variables and UK-SSB cruises. In each 1038 panel, the area above the horizontal dotted line in the left plot (shelf break) is the same as that 1039 to the left of the vertical dotted line in the right plot (on the shelf), and the black points 1040 indicate where samples were collected. Points A, B, CCS and C are located on Fig. 1. (a) and 1041 (b) show $C_{\rm T}$ in winter and summer 2014 respectively, while (c) shows $\Delta p \rm CO_2$ in summer 1042 2014. (d), (e) and (f) show the evolution of DIN from spring through summer to autumn 1043 2014. (g) and (h) illustrate changes in salinity between winter and summer 2015, while (i) 1044 shows A_T in summer 2014. A full set of transects for every variable and UK-SSB cruise is provided in Supp. Figs. S6-S11. 1045



Figure 8. Time-series of regression slopes between S and (a) $C_{\rm T}$, (b) DIN, and (c) SRP. 1050 1051 Negative values indicate that the on-shelf deep layer is enriched in the nutrient relative to the open ocean, while positive values indicate on-shelf depletion. Error bars show the standard 1052 1053 error in the gradient at each point (Table 3). In (b), the predicted relationship based on 1054 dSRP/dS is also shown, considered to represent the component driven by OM 1055 remineralisation. The offset between this and the solid line (mean \pm SD difference = 7.5 \pm 2.1 mmol m⁻³) was likely driven by denitrification/anammox in the Celtic Sea; note that this 1056 1057 offset value is reported per unit practical salinity (which is itself dimensionless). 1058 1059



Figure 9. Changes in (a) $C_{\rm T}$, (b) DIN and (c) SRP, across the entire Celtic Sea transect from winter 2014 (blue) to winter 2015 (red). The × symbols show the mean conditions at CCS each winter. Salinity values increase towards the open ocean. The increase in $C_{\rm T}$ observed at CCS from winter 2014 to winter 2015, and corresponding declines in DIN and SRP, appear to be consistent across much of the Celtic Sea, rather than being a feature of spatial variability local to CCS.

1069 9. Supplementary information

1070 **9.1. Supplementary tables**

1071 Table S1. Summary of the UK-SSB research cruises.

Cruise code ^a	Cruise sampling	Cruise Principle	$C_{\rm T}/A_{\rm T}$ sample	No. of
	dates (dd/mm/yy) ^b	Investigator	collectors	$C_{\mathrm{T}}/A_{\mathrm{T}}$
				samples
DY008	21/03-04/04/14	H. A. Ruhl	L. Darroch	45
JC105	16/06—22/06/14	J. E. Hopkins	J. Fox, R. Houlding,	105
			P. Nelson	
DY026A	04/08—13/08/14	R. Sanders	R. Sims	60
DY026B	17/08—22/08/14	D. Sivyer	R. Sims	16
DY018	10/11-01/12/14	J. Sharples	L. Darroch,	301
			J. E. Hopkins	
DY021	03/03-24/03/15	E. M. S.	N. Hicks	47
		Woodward		
DY029	02/04—28/04/15	A. Poulton	A. Poulton, K. Mayers	214
DY030	05/05-23/05/15	G. Fones	R. Sims	72
DY033	13/07-01/08/15	C. M. Moore	R. Sims	201
DY034	07/08-31/08/15	H. A. Ruhl		52

^aCruise codes beginning with DY were on the RRS *Discovery*, while JC stands for the RRS

1073 James Cook. ^bCruise sampling dates indicate the time period for collection of samples used in

1074 this study, not necessarily the entire duration of the cruises.

1075

1077 Table S2. Coefficients fitted to Eq. (1) at CCS for each time interval (cf. Table 1 and Supp.

1078 Figs. S1–S4) for the variables DIC, DIN, DIP, and practical salinity. Units for v_0 and v_1 are 1079 the same as for the relevant variable (i.e. mmol m⁻³ for C_T , DIN and SRP, and dimensionless 1080 for prostical salinity), while z_1 and z_2 are in m

|--|

		C_{T}				DIN			
Interval code	v_0	<i>v</i> ₁	z_0	z_1	v ₀	<i>v</i> ₁	<i>Z</i> 0	z_1	
W14	2199.3	0	0	0	8.22	0	0	0	
lSp14	2168.9	38.70	40.0	9.64	4.64	4.391	43.8	6.80	
Su14	2173.2	42.87	36.2	8.36	4.68	4.542	39.9	6.48	
A14	2190.3	27.88	54.4	9.99	6.48	3.808	53.3	9.24	
W15	2207.1	-1.08	80.0	12.38	7.34	0.056	28.0	0.69	
IB1	2206.6	2.04	70.7	0.68	6.56	0.529	52.9	15.83	
IB2	2203.6	7.86	35.8	13.93	5.37	1.465	34.3	17.85	
IB3	2192.8	15.97	26.8	19.54	3.21	3.549	19.7	26.09	
IB4	2197.8	12.38	36.5	15.30	4.37	2.385	35.8	18.15	
IB5	2195.1	13.27	42.3	7.71	4.70	2.027	41.7	8.91	
IB6	2195.5	11.54	30.2	13.78	4.06	2.897	32.4	8.04	
IB7	2191.3	17.68	35.9	9.39	3.92	2.944	35.2	10.53	
IB8	2186.6	21.58	37.7	3.21	4.03	3.247	37.2	4.89	
eSp15	2186.4	21.92	35.1	4.04	4.17	3.634	35.6	5.34	
Su15	2180.1	37.27	44.1	12.17	4.20	4.122	48.6	7.78	
		SRI)]	Practical s	alinity		
Interval code	<i>v</i> ₀	<i>v</i> ₁	<i>Z</i> 0	z_1	v ₀	<i>v</i> ₁	<i>Z</i> 0	<i>Z</i> 1	
W14	0.59	0	0	0	35.310	0	0	0	
lSp14	0.37	0.303	42.1	10.80	35.324	-0.0101	18.8	2.33	
Su14	0.36	0.302	38.8	7.01	35.395	0.0429	21.8	17.17	
A14	0.45	0.248	54.2	9.41	35.437	0.0444	58.1	2.45	
W15	0.49	0.040	0.4	29.42	35.356	0.0096	15.0	3.33	
IB1	0.50	0.025	51.6	12.65	35.343	0.0222	54.8	14.72	
IB2	0.43	0.089	36.5	14.87	35.338	-0.0251	23.7	52.69	
IB3	0.27	0.250	17.4	30.97	35.326	-0.0097	43.7	18.90	
IB4	0.34	0.146	37.0	20.29	35.321	-0.0106	31.8	6.99	
IB5	0.38	0.143	42.1	8.16	35.326	-0.0174	42.4	6.72	

IB6	0.34	0.189	33.1	8.33	35.322	-0.0168	34.9	8.36
IB7	0.33	0.184	36.5	11.04	35.312	-0.0058	42.3	0.90
IB8	0.34	36.363	0.16	5.02	35.311	0.0041	35.3	2.01
eSp15	0.35	35.486	0.2	6.03	35.322	0.0218	35.9	7.20
Su15	0.34	48.082	0.27	9.08	35.398	-0.0266	35.1	0.03





Figure S1. Dissolved inorganic carbon (C_T) profiles at each time point at CCS (Table 1). The grey circles show the entire CCS dataset, while the blue circles highlight the data at each time point. The blue line shows the final vertical profile used for inventory calculations (Supp. Table S2).

1103



Figure S2. Dissolved inorganic nitrogen (DIN, nitrate + nitrite + ammonium) profiles at each time point at CCS (Table 1). The grey circles show the entire CCS dataset, while the blue circles show the data at each time point. The blue line shows the final vertical profile used for inventory calculations (Supp. Table S2).



Figure S3. Soluble reactive phosphorus (SRP) profiles at each time point at CCS (Table 1).
The grey circles show the entire CCS dataset, while the blue circles show the data at each
time point. The blue line shows the final vertical profile used for inventory calculations
(Supp. Table S2).





Figure S4. Practical salinity profiles at each time point at CCS (Table 1). The grey circles show the entire CCS dataset, while the blue circles show the data at each time point. The blue line shows the final vertical profile used for inventory calculations (Supp. Table S2).





Figure S5. Seawater pH on the Free scale at each time point at CCS (Table 1), calculated from $A_{\rm T}$ and $C_{\rm T}$ measurements (van Heuven et al., 2011). The grey circles show the entire CCS dataset, while the blue circles show the data at each time point.



Figure S6. Transects of dissolved inorganic carbon (C_T) measured across the Celtic Sea for all of the SSB cruises. In each panel, the area above the horizontal dotted line in the left plot is the same as that to the left of the vertical dotted line in the right plot. Black points indicate the sample locations. The geographical locations of points A, B, CCS and C are shown by Fig. 1.



1216 Figure S7. Transects of Δp CO₂, calculated from C_T and A_T measurements, across the Celtic

1217 Sea for each SSB cruise. The layout is as described for Fig. 7.

1218





Figure S8. Transects of total alkalinity (A_T) across the Celtic Sea for all of the UK-SSB cruises. In each panel, the area above the horizontal dotted line in the left plot is the same as

that to the left of the vertical dotted line in the right plot. Black circles indicate sample

1235 locations.

1236





1250 Figure S9. Transects of dissolved inorganic nitrogen (DIN, nitrate + nitrite + ammonium)

1251 across the Celtic Sea for all of the UK-SSB cruises. In each panel, the area above the

1252 horizontal dotted line in the left plot is the same as that to the left of the vertical dotted line in

1253 the right plot. Black circles indicate sample locations.



Figure S10. Transects of soluble reactive phosphorus (SRP) across the Celtic Sea for all of the UK-SSB cruises. In each panel, the area above the horizontal dotted line in the left plot is the same as that to the left of the vertical dotted line in the right plot. Black circles indicate sample locations.



1285 Figure S11. Transects of practical salinity across the Celtic Sea for all of the UK-SSB cruises.

1286 In each panel, the area above the horizontal dotted line in the left plot is the same as that to

1287 the left of the vertical dotted line in the right plot. Black circles indicate sample locations.

1288

1289



Figure S12. Relationship between dissolved inorganic carbon (C_T) and practical salinity for each UK-SSB cruise. The points used to generate each regression are shown in blue (Section 2.8), and the slope is indicated towards the top right of each panel (see also Fig. 7 and Table 3).



Figure S13. Relationship between dissolved inorganic nitrogen (DIN, nitrate + nitrite + ammonium) and practical salinity for each UK-SSB cruise. The points used to generate each regression are shown in blue (Section 2.8), and the slope is indicated towards the top right of each panel (see also Fig. 7 and Table 3).



Figure S14. Relationship between soluble reactive phosphorus (SRP) and practical salinity
for each UK-SSB cruise. The points used to generate each regression are shown in blue
(Section 2.8), and the slope is indicated towards the top right of each panel (see also Fig. 7)

1341 and Table 3).